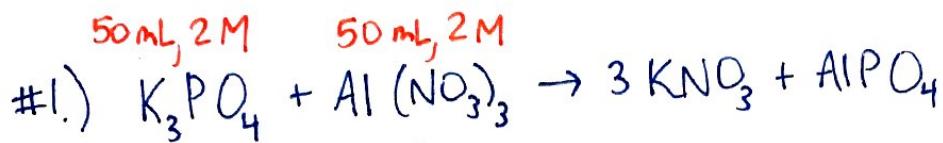


AP Unit 2 Quiz: Review Kahoot



both limiting!

$$50 \text{ mL} \times 2 \text{ M} = 100 \text{ mmol } K_3PO_4 \times \frac{3 \text{ K}^+}{1 \text{ K}_3PO_4} = 300 \text{ mmol K}^+$$

$$[K^+] = \frac{300 \text{ mmol}}{100 \text{ mL}} = 3 \text{ M}$$

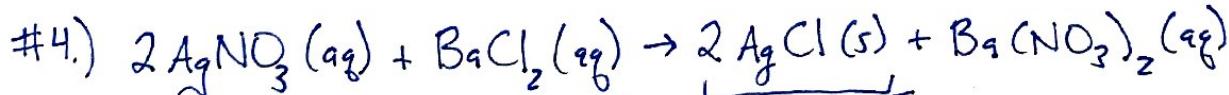
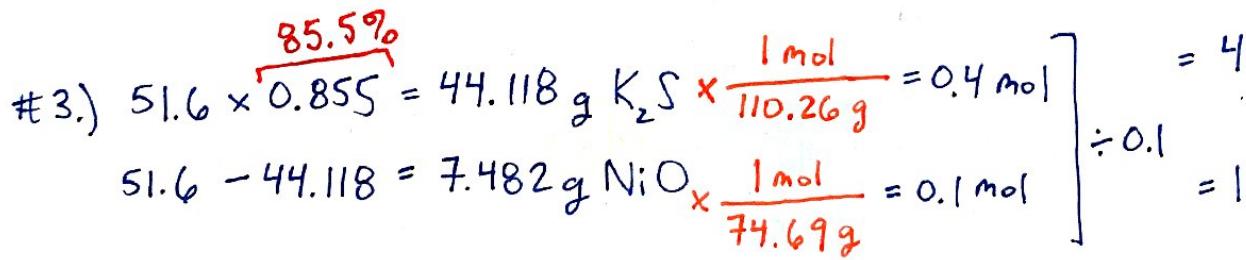
50 mL + 50 mL



$$0.250 \text{ L} \times 5.0 \text{ M} = 1.25 \text{ mol } \text{KClO}_3 \times \frac{3 \text{ O}_2}{2 \text{ KClO}_3} = 1.875 \text{ mol O}_2$$

315 K ↴

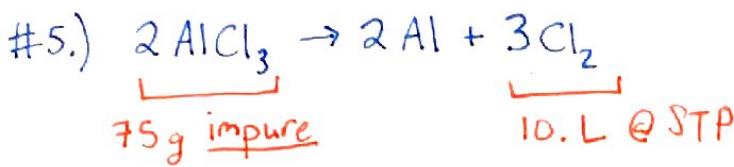
$$V = \frac{nRT}{P} = \frac{(1.875)(62.36)(42+273)}{635} = \boxed{58 \text{ L}}$$



25 mL, 3M 25 mL, 3M

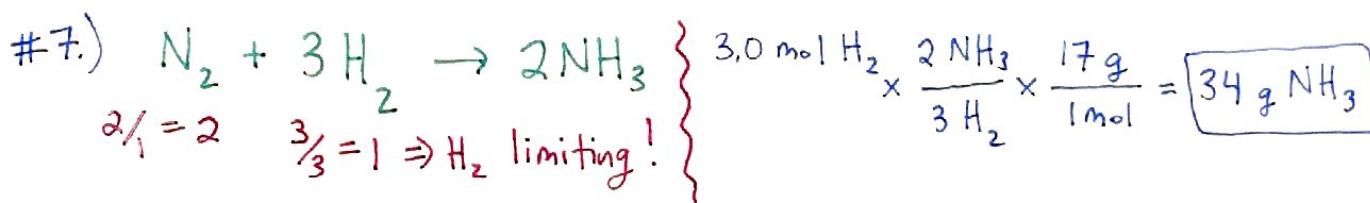
M 25 mL, 3M
2 : 1 \Rightarrow AgNO₃ limiting! only 1 precipitate

* Double replacement rxn \Rightarrow NOT redox



$$10. \text{ L Cl}_2 \times \frac{1 \text{ mol}}{22.4 \text{ L}} \times \frac{2 \text{ AlCl}_3}{3 \text{ Cl}_2} \times \frac{133.3 \text{ g}}{1 \text{ AlCl}_3} = 39.67 \text{ g AlCl}_3$$

$$\% = \frac{39.67 \text{ g}}{75 \text{ g}} \times 100 = \boxed{53\%}$$



$$\frac{29 \text{ g NaCl}}{58 \text{ g/mol}} = 0.5 \cancel{\text{ / }_2} = 0.25 \Rightarrow \text{NaCl limiting} \quad \left. \begin{array}{l} \frac{29 \text{ g CaO}}{56 \text{ g/mol}} \approx 0.5 \cancel{\text{ / }_1} = 0.5 \end{array} \right\}$$

$$29 \text{ g NaCl} \times \frac{1 \text{ mol}}{58 \text{ g}} \times \frac{1 \text{ CaCl}_2}{2 \text{ NaCl}} \times \frac{110.98 \text{ g}}{1 \text{ mol}} = \boxed{28 \text{ g CaCl}_2}$$



$$184 \text{ g Na}_2\text{SO}_4 \times \frac{1 \text{ mol}}{142 \text{ g}} \times \frac{2 \text{ H}_2\text{O}}{1 \text{ Na}_2\text{SO}_4} = \boxed{2.60 \text{ mol H}_2\text{O}}$$



$$\frac{0.31}{1} = 0.31 \quad \frac{0.52}{2} = \underline{0.26} \Rightarrow \text{HCl limiting}$$

$$0.52 \text{ mol HCl} \times \frac{1 \text{ H}_2}{2 \text{ HCl}} \times \frac{2 \text{ g}}{1 \text{ mol}} = \boxed{0.52 \text{ g}}$$